AN EFFICIENT APPROACH TOWARD VIRANTMYCIN: STEREOSPECIFIC CONSTRUCTION OF TETRAHYDROQUINOLINE RING SYSTEM EMPLOYING **INTRAMOLECULAR NITRENE-ADDITION REACTION**

Yoshiki Morimoto, Fuyuhiko Matsuda. and Harubisa Shimhama*

Departmeni of Chemistry. Faculty of Science, Hokkaido University. Sapporo 060. Japan

Summary: An efficient and stereospecific synthesis of the tetrahydroquinoline ring system of the structurally unique alkaloid, virantmycin (1). has been achieved using intramolecular addition reaction of nitrene to olefin as a key step.

The unusual alkaloid, virantmycin **(1).** isolated from the fermentation broth of Streptomyces nitrosporeus by Omura et al. in 1981, has been found to possess antifungal and potent inhibitory activity against various RNA and DNA viruses.¹ Its gross structure has been established as unique tetrahydroquinoline skeleton^{2,3} and the two multi-step syntheses have been reported. $4,5$

In an effort to develop the efficient synthetic scheme for **1. we** found that the tetrahydroquinoline ring can be constructed effectively employing stereospecific intramolecular addition of nitrene, generated by photolysis of aryl azide, 7 to olefin. Although some stereospecific intermolecular nitrene-addition to olefin have been reported.⁸ utilization of the nitrene-addition for organic synthesis has been hitherto restricted because of low yields of the desired aziridines. In this communication we wish to report the stereospecific synthesis of a diastereomeric pair of the tetrahydroquinoline systems 6 and 8 via intramolecular nitreneaddition reaction of the azide-olefins 3 and 4. respectively, and unambiguous assignment for relative configuration of 6 and 8.

The syntheses of azide-olefins 3 and 4 required for nitrene-addition reaction were carried out by stereoselective olefination of the aldehyde 2 easily prepared from 2-allylaniline⁹ (Scheme 1). The Wittig reaction of aldehyde 2 with methyl 2-(triphenylphosphoranylidene) butyrate¹⁰ yielded E-olefin 3 with the stereoselectivity of 32 : 1. On the other hand, the Horner-Emmons reaction of 2 with methyl 2-[bis(2,2,2-trifluoroethyl]phosphono]butyrate¹¹ under Still's condition¹² provided Z-olefin 4 with the stereoselectivity of $8:1$. The stereochemistry of the oleflns 3 and 4 was proven by the chemical shifts of their vinyl protons [S. 6 6.77 (t. J=7.3 Hz]: 4.6 5.94 (t. J=7.3 Hz)] and the presence of NOE shown in 3.

The crucial nitrene-addition reaction by photolysis⁷ of the azide-olefins 3 and 4 proceeded with complete stereospecificity to afford aziridines 5 and 7 in good yields, respectively. The stereochemfstry of the aziridines 5 and 7 was supported from the high field shifts of the protons in the functional groups (CO_2CH_3) or $CH_2CH_3)$ cis to an aromatic ring in their ¹H-NMR spectra [5, CO₂CH₃, δ 3.80 (3H, s), CH₂CH₃, δ 1.50 (1H, dq. J=14.7. 7.3 Hz) and δ 1.20 (1H, dq, J=14.7, 7.3 Hz), CH_2CH_3 , δ 0.84 (3H, t, J=7.3 Hz); 7, CO_2CH_3 , δ 3.25 (3H, s), CH₂CH₃, δ 2.09 (1H, dq, J=14.7, 7.3 Hz) and δ 1.66 (1H, dq, J=14.7, 7.3 Hz), CH₂CH₃, δ 1.04 (3H. t. J=7.3 Hz)]. Reduction of methyl ester, methylation. and the regioselective ring opening of aziridine with inversion¹³ gave two diastereomeric compounds 6 and 8 which contained the same ring system as **1.**

It seemed to be difficult to prove the stereochemistry of compounds 6 and 8 deduced from the stereospecific reaction sequences by their NOE experiments, because these piperidine ring systems were conformationally flexible as shown by the presence of some NOES observed in the NOESY spectra of compounds 8 and 8 (Fig. 1) and ambiguity for the assignment of each proton of 4-H₂ to equatorial or axial orientation remained.¹⁴ In order to fix the conformation of piperidine ring. cyclic carbamates 9 and **10 were** derived from 6 and 8, respectively. As shown in Fig. 2. coupling constants between 3-H and $4-H_2$ in 9 and 10

a. 1) HCI, NaNO₂, then NaN₃ (83%), 2) OsO₄, N-methylmorpholine N-oxide / acetone-H₂O, 3) NaIO₄ / THF-H₂O (63%, 2 steps). b. EtC(=PPh₃)CO₂Me / CH₂Cl₂ / rt / 30 h (80%, 3 : 4 = 32 : 1). c. EtCH[P(O)(OCH₂CF₃)₂]CO₂Me, KN(TMS)₂, 18-crown-6 / THF / -78 °C→rt / 4 h (50%, 3 : 4 = 1 : 8). d. hv / toluene / rt. e. 1) LiAIH₄ / THF / 0 °C, 2) KH / THF, then Mel, 3) Et₄NCI, TFA / CH₂Cl₂ / -15 °C / 20 min. f. 1) AICl₃, ⁿPrSH / CH₂Cl₂, ¹⁵ 2) **Im2C0 (10 equiv.) / toluene.**

Scheme 1.

revealed that these compounds possessed conformationally rigid piperidine ring system. Furthermore, the stereochemistry of compounds 6 and 8 was unambiguously confirmed from NOE experiments with 8 and **10 (Fig.** 2).

In conclusion we have established an efficient and stereospecific route to the tetrahydroquinoline ring system of virantmycin (1) utilizing intramolecular nitrene-addition reaction as a key step. The application of this strategy to the total synthesis of natural product will be reported in due course.

Fig. 1. NOES obsewed in compounds 6 and 8.

Fig. 2. NOES observed in cyclic carbamates 9 and 10.

Acknowledgments: We thank Prof. R A. Raphael and Prof. J. K. M. Sanders (Cambridge University) for the detailed information on the results of their studies and their friendly exchange of views and Dr. H. Naokl (Suntory Institute for Bioorganic Reserch) for kind measurements of NOE spectra.

References and Notes

1. S. Omura. A. Nakagawa. H. Hashimoto. R Oiwa. Y. Iwai. A. Hirano. N. Shibukawa, and Y. Kojima. J. *Antibiotics. 33,* 1395 (1980): A. Nakagawa. Y. Iwai, H. Hasbimoto. N. Miyazaki. R.

Oiwa. Y. Takahashi. A. Hirano. N. Shibukawa. Y. Kojima. and S. Omura. Lbid.. 34. 1408 (1981).

- 2. S. Omura and A. Nakagawa, Tetrahedron Lett., 22, 2199 (1981).
- 3. Stereochemical problem is not settled yet. We have previously suggested that the absolute configuration of **1 is** 2s. 3R through its total synthesis.4 but recently it was reported that the stereochemistry at C-2 was *R by* Sanders et al.6
- 4. Y. Morimoto, K. Oda, H. Shimhama. T. Matsumoto. and S. Omura. Chem Leti. **1988.909.**
- 5. M. L. Hill and R. A. Raphael, Tetrahedron Lett., 27, 1293 (1986).
- 6. C. M. Pearce and J. K. M. Sanders, J. *Chem Sot.. Perkfn 7hns. 1.* **1990, 409.**
- 7. For generation of aryl nitrene by photolysis of aryl azide. see: C. J. Shields, D. R Chrisope. G. B. Shuster. A. J. Dixon. M. Poliakoff. and J. J. Turner. J. *Am Chem Sot.* **109.** 4723 (1987).
- 8. For the stereospecitlc nitrene-addition reaction to olefin. see: R A. Abramovitch and S. R Challand. *J. Chem. Soc., Chem. Commun.*, 1972, 1160; B. V. Ioffe and E. V. Koroleva. *Tetrahedron L&t.* 1973. 619: R A. Abramovitch. S. R Challand. and Y. Yamada. J. Org. Chem. 40. 1541 (1975).
- 9. S. Inoue, N. Takamatsu, and Y. Kishi, *Yakugaku Zasshi*, 97, 553 (1977).
- 10. H. J. Bestmann and H. Schulz, Uebtgs Ann Chem. 674. 11 (1964).
- 11. This phosphonate was prepared from methyl bis(2,2.2-trifluoroethyl)phosphonoacetate by alkylation (NaH, EtI / DMSO / rt / 2 days / 85%). See: J. A. Marshall, B. S. DeHoff, and D. G. Cleary. *J. Org. Chem,* **61.** 1735 (1986).
- 12. W. C. Still and C. Gennarl. *Tetrahedron Leti.* 24. 4405 (1983).
- 13. It is well known that ring opening of aziridine proceed with inversion. See: 0. E. Paris and P. E. Fanta, *J. Am. Chem. Soc.*, **74**, 3007 (1952); N. H. Cromwell, G. V. Hudson, R. A. Wankel, and P. J. Vanderhorst, ibid., 75, 5384 (1953); R. Ghirardelli and H. J. Lucas, ibid., 79. 734 (1957); C. F. Hammer, S. R Heller. and J. H. Craig, *Tetrcheciron,* 28. 239 (1972).
- 14. 1 H-NMR (250 MHz, CDCl₃) data of compounds 6 and 8. 6. 6 7.02 (1H. t. J=7.9 Hz. 7-H). 6.97 (1H. d. J=7.9 Hz, 5-H). 6.68 (1H. t. J=7.9 Hz. 6-H). 6.58 (1H, d, J=7.9 Hz, 8-H), 4.45 (1H, dd, J=6.7, 4.9 Hz, 3-H), 3.41 (2H, s, $13-H₂$), 3.35 (3H, s, 13-OMe), 3.25 (1H, dd, J=17.1, 4.9 Hz, 4-Hs), 3.08 (1H, dd, J=17.1, 6.7 Hz, 4-H₁). 1.63-1.90 (2H. m. 11-H,). 0.94 (3H. t, J=7.3 Hz. 12-H,); 8. 6 7.03 (LH. t. J=7.9 Hz, 7-H). 6.97 (1H. d. J=7.9 Hz, 5-H). 6.67 (1H. t. J=7.9 Hz. 6-H). 6.56 (1H. d. 517.9 Hz, 8-H). 4.35 (lH, dd. J=6.7. 4.9 Hz. 3-H), 4.01 (1H. br s, NH), 3.54 (1H. d. J=9.2 Hz, 13-H). 3.49 (1H. d, J=9.2 Hz, 13-H'), 3.37 (3H, s, 13-OMe), 3.31 (1H, dd, J=17.1, 4.9 Hz, 4-Hs), 3.06 (1H, dd, J=17.1, 6.7 Hz, 4-H₁), 1.58-1.86 (2H, m, 11-H₂), 0.93 (3H, t, J=7.3 Hz, 12-H₃).
- 15. M. Node, K. Nishide. M. Sai. K. Ichikawa. K. Fuji. and E. Fujita, Chem Lett;. **1979. 97:** M. Node, K. Nishide. K. Fuji. and E. Fujita. J. Org. Chem. 45. 4275 (1980).

(Received in Japan 9 July 1990)